Bridging the Gap

The High School–University
Transition in Chemistry

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Chemical Education Survey:

- Pilot study in 2006-7
- First major survey in 2007-8
- Continuing this year...

What factors contribute to a successful high school–university transition?

What can schools and universities do to help students manage this transition?

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Related US Study:

Survey of 12 US colleges & universities:

- R. H. Tai, P. M. Sadler, and J. F. Loehr
 J. Res. Sci. Teaching, 2005, 42(9), 987-1012
- R. H. Tai, R. B. Ward, and P. M. Sadler
 J. Chem. Ed., 2006, 83(11), 1703-1711
- R. H. Tai and P. M. Sadler

 J. Chem. Ed., 2007, 84(6), 1040-1046

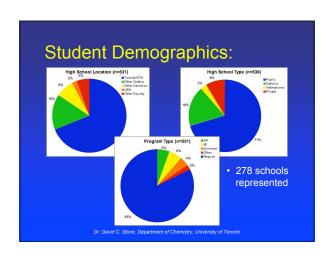
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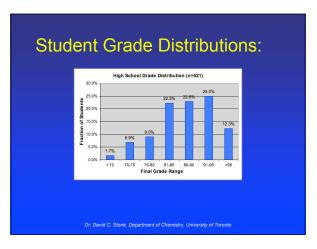
Who Are Our Students?

- 1003 in CHM 138F (Intro. Organic Chem)
- 640 in CHM 139F (Gen. Physical Chem)
- 160 in CHM 151Y (Advanced Intro. Chem)

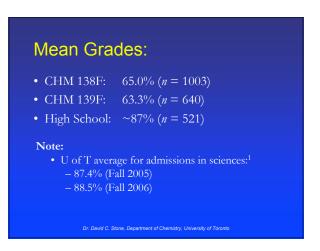
Survey Response Rate ($n = 536$):	29.7%
Female:	60.6%
Male:	39.4%
Native English-speaker:	44.8%
Semestered Courses:	58.3%
Performed Independent Study Unit:	57.7%

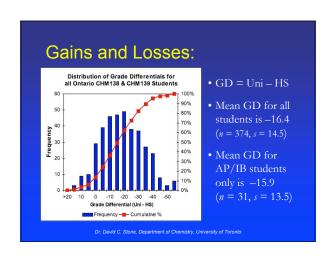
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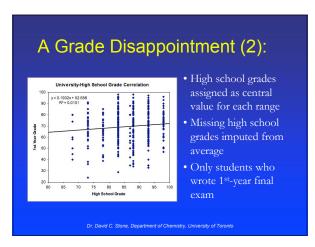


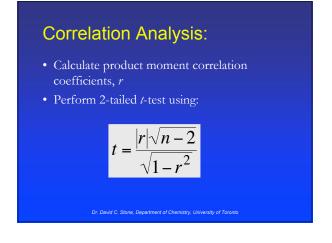


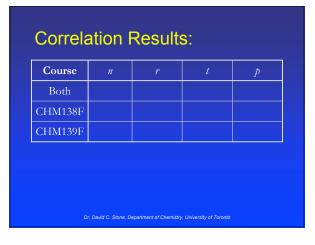




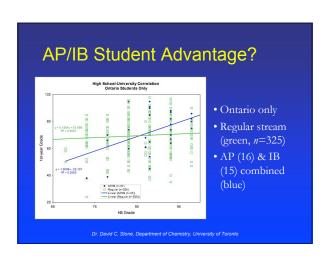








Correlation Results: Both r t p Both 478 0.100 2.193 0.029 CHM138F 333 0.0985 1.801 0.073 CHM139F 145 0.0872 1.046 0.297 Dr. Devid C. Stone, Department of Chemistry, University of Toronto



Correlation Results:

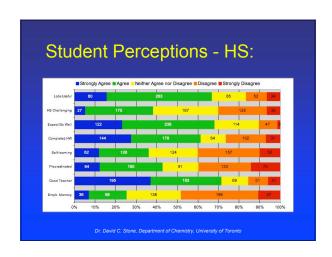
Course	n	r	t	Þ
Both	478	0.100	2.193	0.029
CHM138F	333	0.0985	1.801	0.073
CHM139F	145	0.0872	1.046	0.297
Regular	326	0.0755	1.363	0.174
AP/IB	31	0.4542	2.745	0.010
Enriched	18	-0.2883	1.204	0.246

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Student Perceptions - HS:

- 1. Labs were useful and relevant to topics
- 2. I found HS chemistry challenging
- 3. I expect to do well in university chemistry
- 4. I always completed my homework
- 5. I learnt much independently
- 6. I procrastinated over my homework
- 7. My HS teacher performed effectively
- 8. HS placed more emphasis on memorization

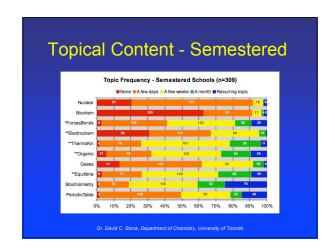
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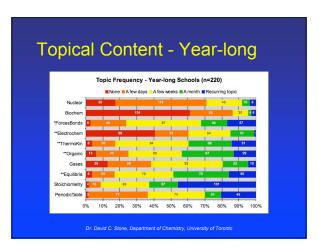


Topical Content:

- Nuclear Chemistry (isotopes, radio decay, etc.)
- Biochemistry (enzymes, proteins, DNA/RNA)
- Forces & Bonding (VSEPR. van der Waal's, etc.)
- Floaterachamieter (rador, calvania & raltain calla
- Thermochemistry & Kinetics (Hess' Law. etc.
- Organic Chemistry (naming, groups, reactions)
- Gases (properties, gas laws)
- Equilibria (reactions, acid/base, solubility)
- Stoichiometry (chemical reactions & equations)
- The Periodic Table (electron config., periodicity)

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Focus Group Themes:

- Varying background preparation
 - e.g. hybridization, stereochemistry, organic reactions
- Test & evaluation practices
 - Problems (exercises) vs. problem-solving
 - calculations vs. application/understanding
- Instructor transitions
 - teaching style, test/exam question format, etc.
- TAs
 - tutorial content/approach
 - lab evaluation, marking, feedback

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References:

- Faculty of Arts & Science Annual Report, July 2006–June 2007, page 9.
- 2. Faculty of Arts & Science Calendar 2008-2009, page 495.

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Correlation by School (1):

School	n	r	t	Þ
A	5	-0.428	0.82	0.472
В	6	-0.732	2.15	0.098
С	7	-0.136	0.31	0.771
D	9	+0.143	0.38	0.714
Е	5	-0.495	0.99	0.397
F	8	+0.172	0.43	0.683
G	7	-0.055	0.12	0.907
Н	5	+0.363	0.67	0.549
J	6	-0.122	0.25	0.818

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Correlation by School (2):

School	n		t	Þ
K	5	-0.062	0.11	0.922
L	9	+0.457	1.36	0.216
M	11	-0.314	0.99	0.347
N	5	+0.612	1.34	0.273
P	5	+0.561	1.17	0.326
Q	5	+0.430	0.82	0.470
R	7	+0.483	1.23	0.272
S	6	+0.496	1.14	0.317
Т	5	-0.704	1.72	0.184

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